



Tropical mangrove sediments as a natural inoculum for efficient electroactive biofilms

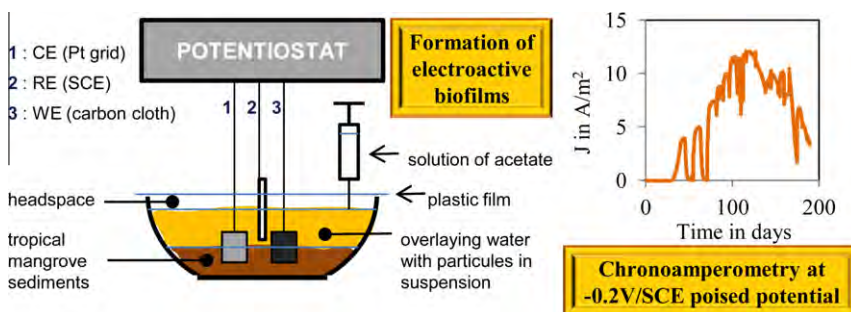
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HIGHLIGHTS

- ▶ Tropical mangrove sediments investigated as a natural inoculum for EA biofilm.
- ▶ A maximal current density of 12 A/m² obtained with an adequate procedure.
- ▶ An enriched medium makes it possible to form secondary EAB more quickly efficient.

GRAPHICAL ABSTRACT



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ABSTRACT

Chronoamperometry is known to be an efficient way to form electroactive biofilms (EAB) on conductive electrodes. For the first time, tropical mangrove sediments are analyzed as a potential inoculum to form MFC anodes with the use of acetate as substrate. The performance of the EAB-coated carbon cloth electrodes are evaluated according to the maximal current density, the coulombic efficiency and the cyclic voltammograms. Working electrodes (WE) polarized at -0.2 V/SCE gave better results compared to -0.4 V/SCE and 0.0 V/SCE. The maximal current density attained was 12 A/m² with a CE of 24%. Contributions of the EAB in the generation of current were discussed and mechanisms of electronic transfer by the bacteria were discussed. Epifluorescence and SEM images showed the evolution of the biofilms on the electrode surface and the heterogeneity of the structure.

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1. Introduction

For a few years now, the technology of the microbial fuel cell (MFC), exploiting the properties of the specific biofilms called electroactive biofilms (EAB), has been of great interest as a future alternative in energy production. The MFC is based on the classical fuel cell principle, with for main difference the catalysis of the electrochemical reactions realized by the EAB, adhered on the electrodes (Kim et al., 1999; Rabaey and Verstraete, 2005; Logan et al., 2006). In general, in the published studies, only the anode is covered with an EAB and the cathode is an abiotic one such as

for example an air-cathode using platinum as a catalyst or the supply of chemical oxidizers (ferricyanide or Mn(IV)) (Rabaey et al., 2003; Rhoads et al., 2005; Shantaram et al., 2005). With more details, the anodic EAB catalyzes the oxidation of the organic matter or the carbon source present in the anodic chamber and transfers the generated electrons to the conductive anode. Then, the difference of potentials between the anode and the cathode induces that the electrons pass through an external circuit composed of generally one resistive device and go to the cathode where the reduction of the dioxygen or another oxidizer takes place abiotically (Logan et al., 2006). A proton exchange membrane separates the anodic and cathodic electrolytes. This membrane constitutes a filter through which the protons from the anodic oxidation can migrate to the cathode to participate to the reduction, but also

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