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# Atomic-scale interactions of the interface between chitosan and Fe<sub>3</sub>O<sub>4</sub>

Linhui Qiang<sup>a</sup>, Zhanfeng Li<sup>a</sup>, Tianqi Zhao<sup>a</sup>, Shuangling Zhong<sup>b</sup>, Hongyan Wang<sup>a</sup>, Xuejun Cui<sup>a,\*</sup>

<sup>a</sup> College of Chemistry, Jilin University, Changchun, 130012, PR China

<sup>b</sup> College of Resources and Environment, Jilin Agricultural University, Changchun, 130118, PR China

### HIGHLIGHTS

- ► We built the interface model between chitosan and Fe<sub>3</sub>O<sub>4</sub>.
- The (111) interface has strongest interaction than other interfaces.
- ► There was high probability of formed hydrogen bonds of (111) interface.
- ► There were interactions between Fe<sub>3</sub>O<sub>4</sub> and nitrogen atoms of chitosan.

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### 1. Introduction

The iron oxide nanoparticles (Fe<sub>3</sub>O<sub>4</sub> NPs) were a kind of important magnetic material. It has been widely used in bio-separation, hyperthermia and magnetic guided drug targeting [1–4]. For biomedical applications, Fe<sub>3</sub>O<sub>4</sub> NPs were often treated by surface modification which could increase the functionality and improve the biocompatibility of the Fe<sub>3</sub>O<sub>4</sub> NPs [5]. Chitosan obtained by deacetylation of chitin was a unique natural linear cationicpolymer, which was the structural element in the exoskeleton of crustaceans

# GRAPHICAL ABSTRACT



## ABSTRACT

Molecular dynamics (MD) simulation was employed to study chitosan adsorption on different  $Fe_3O_4$  crystallographic planes at the atomic level. The interaction energy between chitosan and different  $Fe_3O_4$  surfaces indicates that the interaction of chitosan and  $Fe_3O_4$  (111) surfaces is stronger than that of (110) and (001) surfaces. The concentration profiles show that hydrogen and amino groups of chitosan could form strong interactions with the surfaces of  $Fe_3O_4$ . The radial distribution function show that the probability of forming hydrogen bonds between groups of chitosan and oxygen atoms on  $Fe_3O_4$  (111) surface is more than that of  $Fe_3O_4$  (110) surface, and the nitrogen atoms and iron atoms on the surface have weak physical interactions. This study provides useful information in understanding the interfacial interaction mechanism at the atomistic scale for polymer and mineral.

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and cell walls of fungi. Chitosan had many beneficial properties such as a low toxicity, low immunogenicity, excellent biocompatibility, biodegradability and as well as a high positive charge density. Due to its high positive charge density, it could easily form polyelectrolyte complexes with negatively charged nucleotides or drugs by electrostatic interaction [6–8].

Recently, chitosan modified magnetic particles (MPs) have gained significant attention for biomedical applications. Li et al. has reported the MPs can provide excellent biocompatility, biodegradability and low toxicity without compromising their magnetic targeting [9]. Sun et al. have prepared magnetic targeting chitosan nanoparticles as a drug delivery system and imaging agents for photodynamic therapy to cancer treatment [10]. Nguyen and co-workers have developed a novel chitosan Fe<sub>3</sub>O<sub>4</sub> nanobiocomposite-based platform for electrochemical detection of HIV-1 [11]. Although, MPs have applied in many fields, the

<sup>\*</sup> Corresponding author at: College of Chemistry, Jilin University, Qianjin Street 2699#, Changchun, 130012, PR China. Tel.: +86 431 85168470; fax: +86 431 85168470.

IX. +60 451 65106470.

E-mail address: cui\_xj@jlu.edu.cn (X. Cui).

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