



## Rational design of monolayers for improved water evaporation mitigation

Emma L. Prime<sup>a</sup>, Diana N.H. Tran<sup>a</sup>, Michael Plazzer<sup>b</sup>, Devi Sunartio<sup>a</sup>, Andy H.M. Leung<sup>a</sup>, George Yiapanis<sup>b</sup>, Svetlana Baoukina<sup>b,1</sup>, Irene Yarovsky<sup>b,\*\*</sup>, Greg G. Qiao<sup>a,\*</sup>, David H. Solomon<sup>a,\*</sup>

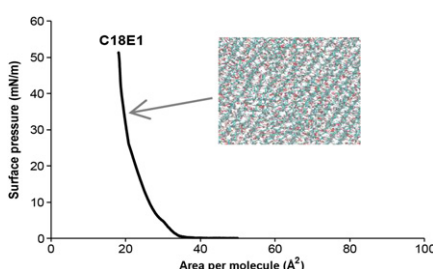
<sup>a</sup> Department of Chemical and Biomolecular Engineering, University of Melbourne, Parkville, VIC 3010, Australia

<sup>b</sup> School of Aerospace, Mechanical and Manufacturing Engineering, RMIT University, Melbourne, VIC 3001, Australia

### HIGHLIGHTS

- ▶ A series of monolayer compounds was synthesized.
- ▶ Their performance as monolayers was investigated under static and dynamic conditions.
- ▶ Molecular simulation used to assist in interpretation of experimental results.
- ▶ Increasing the hydrophilic head-group with one ethylene oxy improved monolayer lifetime.
- ▶ Design principles for future engineering of improved suppressants were suggested.

### GRAPHICAL ABSTRACT



### ARTICLE INFO

#### Article history:

Received 3 July 2012

Received in revised form 6 September 2012

Accepted 21 September 2012

Available online 27 September 2012

#### Keywords:

Monolayer

Water evaporation

Molecular dynamics simulation

Wind resistance

1-Octadecanol

Ethylene glycol monooctadecyl ether

### ABSTRACT

Seven chemically designed monolayer compounds were synthesized and investigated with comparison to the properties and water evaporation suppression ability of 1-hexadecanol and 1-octadecanol. Increasing the molecular weight and polarity of the compound headgroup drastically altered the characteristics and performance of the monolayer at the air/water interface. Contrary to the common expectation the monolayer's lifetime on the water surface decreased with increasing number of ethylene oxy moieties, thus optimal performance for water evaporation suppression was achieved when only one ethylene oxy moiety was used. Replacing the hydroxyl headgroup with a methyl group and with multiple ethylene oxy moieties resulted in a loss of suppression capability, while an additional hydroxyl group provided a molecule with limited performance against water evaporation. Theoretical molecular simulation demonstrated that for exceptional performance, a candidate needs to possess a high equilibrium spreading pressure, the ability to sustain a highly ordered monolayer with a stable isotherm curve, and low tilt angle over the full studied range of surface pressures by simultaneously maintaining H-bonding to the water surface and between the monolayer chains.

© 2012 Elsevier B.V. All rights reserved.

## 1. Introduction

Monolayers are one-molecule thick surface films formed by certain amphiphilic molecules such as fatty alcohols and their derivatives. They have been used as the basis for a number of different applications, such as, membranes for molecular separations [1,2], biomedical systems for tissue engineering and drug delivery [3–5], packaging and coating materials [6,7], and water evaporation mitigation [8]. Water is one of the scarcest resources and it is

\* Corresponding authors. Tel.: +61 3 8344 8665; fax: +61 3 8344 4153.

\*\* Corresponding author. Tel.: +61 3 9925 2571; fax: +61 3 9925 5290.

E-mail addresses: [irene.yarovsky@rmit.edu.au](mailto:irene.yarovsky@rmit.edu.au) (I. Yarovsky),

[gregghq@unimelb.edu.au](mailto:gregghq@unimelb.edu.au) (G.G. Qiao), [davids@unimelb.edu.au](mailto:davids@unimelb.edu.au) (D.H. Solomon).

<sup>1</sup> Visiting scholar from Department of Biological Sciences, University of Calgary, 2500 University Drive NW, Calgary, AB, Canada.