



Pretreatment of wheat straw using SO₂ dissolved in hot water

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HIGHLIGHTS

- ▶ Wheat straw hemicellulose was efficiently pretreated in an H₂O–SO₂ system.
- ▶ SO₂ in the H₂O–SO₂ system could be recovered and reused after reaction.
- ▶ SO₂ in water favors hemicellulose conversion rather than xylose dehydration.
- ▶ Kinetic analysis was performed by Saeman modeling and Arrhenius equation.

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ABSTRACT

Efficient pretreatment is important for complete enzymatic conversion of lignocellulosic materials. Pretreatment of wheat straw with sulfur dioxide (SO₂) dissolved in hot water achieved xylose and total product yields of up to 61.1% and 93.9%, respectively, based on the mass of lignocellulose in wheat straw. The apparent activation energies for hemicellulose conversion and xylose dehydration were 7.8 and 9.0 kJ/mol. FT-IR spectra of the residual solid after treatment showed that the hemicellulosic components were converted, the hydrogen bonds in cellulose were broken, but the lignin structure was not changed. Importantly, the SO₂ was recovered from the product mixture by steam stripping and could be reused. Thus, the SO₂–H₂O system is an efficient and environmentally friendly way for the conversion of hemicellulose in wheat straw into monosaccharides, such as xylose, glucose and arabinose.

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1. Introduction

The utilization of lignocellulosic materials for the production of value-added chemicals or biofuels generally requires a pretreatment process (Dadi et al., 2006; Grous et al., 1986; Kumar et al., 2009; Qi et al., 2009). Chemical pretreatment processes such as dilute acid hydrolysis are efficient, but create waste, require additional recovery steps and can lead to degradation of products. When sulfuric acid is used, the derived sugars, mainly xylose, are easily converted into furfural, which has an inhibitory effect on the subsequent enzymatic hydrolysis (Almeida et al., 2009).

Orozco et al. (2007) reported that H₃PO₄, a medium strong acid, could promote the yield of glucose from cellulose, since its acidity is just sufficient to break the bonds of cellulose, but not too strong to control the reaction process. SO₂ dissolved in water can react with water to form sulfurous acid as a medium strong acid, like H₃PO₄, to generate H⁺ (Donaldson et al., 2009) and also can exist as a Lewis acid (Jonas et al., 1994). Thus, SO₂ in water may potentially be able to convert hemicellulose with a high sugar yield, but

a low yield of furfural. Moreover, SO₂ can be removed by steam stripping from an aqueous solution, and the recovered SO₂ can be reused. However, pretreatment of wheat straw in an H₂O–SO₂ system has not been reported in the literature. Therefore, in the present study wheat straw was treated in hot water using SO₂ as a recoverable catalyst. Kinetics analysis of the hemicellulose conversion and xylose degradation with time were established, and the reaction rate constants and apparent activation energy were calculated according to Saeman models and Arrhenius equation.

2. Experimental

2.1. Chemicals

Wheat straw was dried at 105 °C for 12 h and milled to small particles (less than 0.5 mm). The contents of cellulose and hemicellulose in the wheat straw were 40.7% and 30.1%, respectively, as measured with an ANKOM A2000i automatic fiber analyzer. Aqueous 6% solutions of SO₂ were purchased from Aladdin Reagent Inc. (Shanghai, China). N₂ with a volume fraction purity of 0.9995 was supplied by Beijing Haipu Gases Ltd. (Beijing, China). Anhydrous D-glucose, xylose, arabinose, furfural and levulinic acid (LA) with

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