



Preparation of oxygen-vacant TiO_{2-x} and activated carbon fiber composite using a single-step thermal plasma method for low-concentration elemental mercury removal

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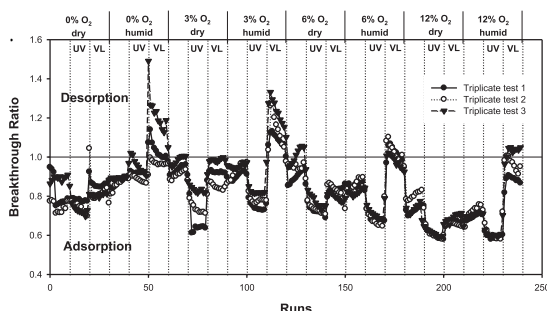
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HIGHLIGHTS

- ▶ TiO_{2-x} /ACF composites were developed via a $\text{N}_2/\text{Ar}/\text{He}$ thermal plasma system.
- ▶ TiO_{2-x} had a size within 10–40 nm and a mixture of anatase and rutile phases.
- ▶ TiO_{2-x} /ACF composites had a greater Hg removal under UV than VL irradiation.
- ▶ Presence of O_2 enhanced the Hg removal of TiO_{2-x} /ACF.
- ▶ Moisture reduced Hg removal performance of TiO_{2-x} /ACF.

GRAPHICAL ABSTRACT

TiO_{2-x} /ACF composites were synthesized using $\text{N}_2/\text{He}/\text{Ar}$ thermal plasma. Hg^0 removal by the TiO_{2-x} /ACF was enhanced by light radiation and O_2 but reduced by competitive adsorption from moisture.



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ABSTRACT

Oxygen-vacant TiO_2 (TiO_{2-x}) nanoparticles and TiO_{2-x} /activated carbon fiber (ACF) composites were developed via a $\text{N}_2/\text{Ar}/\text{He}$ thermal plasma system. The TiO_{2-x} nanoparticles and TiO_{2-x} /ACF composites were characterized with TEM, XRD, UV–Vis, ESEM and N_2 adsorption isotherms. The removal effectiveness of TiO_{2-x} /ACF for gaseous Hg^0 at ppb concentration level and various conditions was subsequently evaluated. The experimental results indicated that the formed TiO_{2-x} nanoparticles had a size within 10–40 nm and a mixture of anatase and rutile phases. The TiO_{2-x} formed at 7% N_2 concentration had an evident red-shift in wavelength absorption. The ESEM and N_2 adsorption results suggested that the synthesized TiO_{2-x} nanoparticles unevenly deposited on the ACF surface causing a decrease in total and micropore surface areas/volumes. Hg breakthrough tests revealed that TiO_{2-x} /ACF composites had a greater Hg removal under UV or visible-light irradiation than those obtained in the dark condition. The presence of O_2 up to 12% greatly enhanced the Hg removal, implying the positive effects of catalytic oxidation. However, moisture reduced Hg removal performance, especially when visible-light irradiation was applied. These results revealed the competitive adsorption between Hg species and H_2O and the physisorption nature of Hg species on the light-induced hydrophilic TiO_{2-x} /ACF surface.

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