



Catalytic decomposition of 4-phenoxyphenol to aromatics over Pd/XCs_{2.5}H_{0.5}PW₁₂O₄₀/ACA (X = 10, 20, 30, 40, and 50 wt%) catalysts

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ABSTRACT

Activated carbon aerogel (ACA) was prepared by a chemical activation of carbon aerogel using phosphoric acid (H₃PO₄). A series of Cs_{2.5}H_{0.5}PW₁₂O₄₀-impregnated activated carbon aerogels (XCs_{2.5}H_{0.5}PW₁₂O₄₀/ACA) were prepared with a variation of Cs_{2.5}H_{0.5}PW₁₂O₄₀ content (X) in order to provide acid sites to ACA. Palladium catalysts were then supported on Cs_{2.5}H_{0.5}PW₁₂O₄₀-impregnated activated carbon aerogel (Pd/XCs_{2.5}H_{0.5}PW₁₂O₄₀/ACA, X = 10, 20, 30, 40, and 50 wt%) by an incipient wetness impregnation method for use in the decomposition of 4-phenoxyphenol to aromatics. 4-Phenoxyphenol was used as a lignin model compound for representing 4–O–5 linkage of lignin. Cyclohexanol, benzene, and phenol were mainly produced by the decomposition of 4-phenoxyphenol. Conversion of 4-phenoxyphenol and total yield for main products (cyclohexanol, benzene, and phenol) were closely related to the acidity of Pd/XCs_{2.5}H_{0.5}PW₁₂O₄₀/ACA. Conversion of 4-phenoxyphenol and total yield for main products increased with increasing acidity of Pd/XCs_{2.5}H_{0.5}PW₁₂O₄₀/ACA. Among the catalysts tested, Pd/20Cs_{2.5}H_{0.5}PW₁₂O₄₀/ACA with the largest acidity showed the highest conversion of 4-phenoxyphenol and total yield for main products. Conversion of 4-phenoxyphenol and total yield for main products over Pd/20Cs_{2.5}H_{0.5}PW₁₂O₄₀/ACA were much higher than those over palladium catalyst supported on commercial activated carbon (Pd/AC) and palladium catalyst supported on activated carbon aerogel (Pd/ACA).

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1. Introduction

Carbon dioxide from burning of fossil fuels has been considered as a contributor to global warming. Furthermore, fossil fuels such as coal and crude oil are not sustainable [1–3]. Biomass has attracted much attention as an environmentally friendly and renewable energy source, because biofuels can reduce the dependence on fossil fuel and the emission of carbon dioxide [4].

Corn sugar and vegetable oil, which are known as first generation biofuels, are environmentally friendly, but feedstocks for first generation biofuels are limited and compete with food resource [5,6]. However, second generation biofuels derived from lignocellulosic feedstocks do not compete with food resource. Lignocellulose is typically composed of three main components: cellulose, hemicellulose, and lignin. Lignin produced by delignification process in the pulp industries is mainly burned as a low value fuel for power generation [7,8]. Therefore, decomposition of lignin to aromatics becomes as an advantageous technology for producing aromatics as

biofuels, because lignin is the richest source of aromatics in nature [9,10].

Many attempts have been made on the catalytic decomposition of lignin to aromatics, because thermal cracking of lignin requires high temperature (over 450 °C) and pressure (over 100 atm) for considerable performance [11–14]. Liquid acid such as H₃PO₄ and solid acid such as cesium-exchanged heteropolyacid (Cs_xH_{3.0–x}PW₁₂O₄₀, x = 2.0–3.0) are known as promising catalysts for the decomposition of lignin to aromatics [15–17]. Novel metal catalysts supported on carbon (Pd/C, Pt/C, and Rh/C) also show a considerable performance for selective decomposition of C–O bond in lignin [18,19].

Dimeric lignin model compounds for representing C–O and C–C bonds in lignin have been used as a lignin feedstock due to the complex structure of lignin [20–22]. In particular, dimeric chemical compounds containing C–O bond such as β–O–4, α–O–5, and 4–O–5 have been used as lignin model compounds, because C–O bond is abundant linkage type in the lignin. Among various lignin model compounds, 4-phenoxyphenol has been widely employed as a lignin model compound for representing 4–O–5 bond in lignin [23,24].

Carbon aerogel (CA) [25–29] and activated carbon aerogel (ACA) [30–35] have been employed as catalyst support, electrochemical

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