

Magnetic chains formed from tetra-coordinate Co(II) complexes[‡]

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Two tetra-coordinate Co(II) complexes, $[\text{CoCl}_2(\text{bzfupy})_2]$ and $[\text{CoCl}_2(\text{p-tol})_2]$ with N-donor ligands, are apparently mononuclear systems. The magnetic susceptibility at low temperature attains a maximum which is a fingerprint of the exchange interaction. The exchange coupling constant fitted to a model of 1D-magnetic chain yields $J/hc = -0.50 \text{ cm}^{-1}$ and -0.57 cm^{-1} , respectively. The magnetic interaction is mediated by the $\pi-\pi$ stacking in the first complex, and the short Cl···Cl contacts in the second.

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Introduction

Cobalt(II) complexes are very variable from the perspective of magnetochemistry (Lloret et al., 2008; Papánková et al., 2010). They adopt either a quasioctahedral hexa-coordinate configuration, or a quasitetrahedral tetra-coordinate configuration, although penta-coordination is also frequent.

In the compressed tetragonal bipyramid, the ground electronic term is ${}^{4}A_{1g}$ and this is subjected to zero-field splitting. As a consequence, the effective magnetic moment on cooling from the laboratory temperature remains constant and it decreases at low temperature. The zero-field splitting is a gap between the two lowest crystal-field multiplets (Kramers doublets) $\Delta = 2D = E(\Gamma_7) - E(\Gamma_6) > 0$. The reported D-values are high: $D \sim 10^2 \text{ cm}^{-1}$ (Boča, 2004; Titiš & Boča, 2011). By contrast, for the elongated tetragonal bipyramid, the ground electronic term is orbitally degenerate $-{}^{4}E_{g}$ and it produces four Kramers doublets; this situation needs modelling beyond the spin-Hamiltonian formalism (Boča, 2006). The effective magnetic moment, on cooling, then passes through a round maximum.

In tetra-coordinate complexes, the ground term is ${}^{4}A_{2}$ and the situation can be described by the spin Hamiltonian that accounts for the zero-field splitting. Moderate *D*-values, both negative and positive, were reported (Krzystek et al., 2004); in the regular tetrahedron D = 0 holds true.

A lot of dinuclear and trinuclear complexes are known where the *D*-parameter has been determined by fitting the magnetic data (Hudák et al., 2011).

In this article we report on two mononuclear Co(II) complexes of the general formula $[CoCl_2L_2]$ where L stands for an *N*-donor heterocyclic ligand: bzfupy = benzofuro[3,2-*c*]pyridine and p-tol = *p*-toluidine.

Experimental

 $CoCl_2 \cdot 6H_2O$, toluidine, and ethanol were purchased from commercial sources and used as received. The target complexes $[CoCl_2(bzfupy)_2]$ and $[CoCl_2(p$ $tol)_2]$, subsequently *I* and *II*, were prepared by the methods already described elsewhere (Vrábel et al., 2007; Malinovskii, 1957).

Electronic spectra (in Nujol mull) were measured using a Specord 200 instrument (Analytik Jena AG,

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